Access DB# 18493

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Inventors (please provide full names):	Shin U	itsunomi	19 + Serio	
<u> Vamada</u>)	
Earliest Priority Filing Date:	F01/16/2	∞	\ ,	
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Date Searcher Picked Up:	Bibliographic Litigation			_
Searcher Prep & Review Time:	Fulltext	•		
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Online Time:	Other	Other (specify)		

PTO-1590 (8-01)

=> file reg
FILE 'REGISTRY' ENTERED AT 16:16:04 ON 30 OCT 2002
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STRUCTURE FILE UPDATES: 29 OCT 2002 HIGHEST RN 467418-81-1 DICTIONARY FILE UPDATES: 29 OCT 2002 HIGHEST RN 467418-81-1

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(FILE 'HOME' ENTERED AT 15:42:24 ON 30 OCT 2002)

FILE 'LREGISTRY' ENTERED AT 15:43:59 ON 30 OCT 2002

L1 STR L2 STR

L3 STR

FILE 'REGISTRY' ENTERED AT 16:02:27 ON 30 OCT 2002

L4 SCR 2043

L5 1 S L1 AND L2 AND L3 AND L4

L6 12 S L1 AND L2 AND L3 AND L4 FUL

SAV L6 MCC025/A

FILE 'CAOLD' ENTERED AT 16:08:06 ON 30 OCT 2002

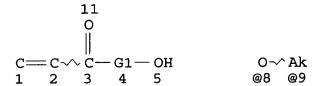
L7 0 S L6

FILE 'ZCAPLUS' ENTERED AT 16:08:07 ON 30 OCT 2002 L8 10 S L6

FILE 'REGISTRY' ENTERED AT 16:16:04 ON 30 OCT 2002

=> d 16 que stat

L1 STR



REP G1=(1-10) 8-3 9-5
NODE ATTRIBUTES:
CONNECT IS E2 RC AT 9
DEFAULT MLEVEL IS ATOM
GGCAT IS SAT AT 9
DEFAULT ECLEVEL IS LIMITED
ECOUNT IS M2-X3 C AT 9

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 8

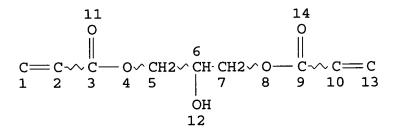
STEREO ATTRIBUTES: NONE L2 STR

 $C = C \sim COOH$ 1 2 3

NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 3

STEREO ATTRIBUTES: NONE L3 STR



NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE L4 SCR 2043

L6 12 SEA FILE=REGISTRY SSS FUL L1 AND L2 AND L3 AND L4

100.0% PROCESSED 3693 ITERATIONS

12 ANSWERS

SEARCH TIME: 00.00.08

=> file zcaplus FILE 'ZCAPLUS' ENTERED AT 16:16:19 ON 30 OCT 2002 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2002 AMERICAN CHEMICAL SOCIETY (ACS)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

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=> d 18 1-10 cbib abs hitstr hitrn

L8 ANSWER 1 OF 10 ZCAPLUS COPYRIGHT 2002 ACS
2000:865391 Document No. 134:30468 Photocurable polyisocyanate-based composition for coating. Matoba, Takao; Suzuki, Kenya; Yushima, Hajime (Kansai Paint Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 2000344856 A2 20001212, 9 pp. (Japanese). CODEN: JKXXAF. APPLICATION: JP 1999-162173 19990609.

$$(C_2H_5)_2N$$
 $C = CH - CH = CH - C$
 $N(C_2H_5)_2$
 $N(C_2H_5)_2$
 $N(C_2H_5)_2$

The compn. comprises (A) a compd. having radical polymerizable unsatd. and active hydrogen groups, (B) a photoinitiator of visible photopolymn. initiator having max. absorption wave length in the visible optical territory and/or a photoinitiator having max. absorption wave length in the near IR radiation territory, (C) a photopolymn. initiator having max. absorption wave length in the UV ray territory, (D) a polyisocyanate. Thus, reaction acrylic acid soln. contg. tetrabutylammonium chloride and p-benzoquinone and a copolymer of styrene, MMA, iso-Bu methacrylate, Bu acrylate, qlycidyl methacrylate, and 2-hydroxyethyl acrylate, and mixing with 2-hydroxy-3-acryloxyoxypropyl methacrylate, TPA 100 (HDI polyisocyanate), I, Irgacure 819, di-Et thioxanthone, dibutyltin dilaurate gave a coating compn.

IT 311346-48-2

(photocurable polyisocyanate-based compn. for coating)

RN 311346-48-2 ZCAPLUS

CN 2-Propenoic acid, 2-methyl-, butyl ester, polymer with Duranate TPA 100, ethenylbenzene, 2-hydroxyethyl 2-propenoate, 2-hydroxy-3-[(1-oxo-2-propenyl)oxy]propyl 2-methyl-2-propenoate, methyl 2-methyl-2-propenoate, 2-methylpropyl 2-methyl-2-propenoate and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 134498-50-3 CMF Unspecified CCI PMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 1709-71-3 CMF C10 H14 O5

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{--CH}_2\text{--O-C-CH} \end{array}$$

CM 4

CRN 100-42-5 CMF C8 H8

$$H_2C = CH - Ph$$

CM 5

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 6

CRN 97-86-9 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{i-BuO-C-C-Me} \end{array}$$

CRN 80-62-6 CMF C5 H8 O2

CM 8

CRN 79-10-7 CMF C3 H4 O2

IT 311346-48-2

(photocurable polyisocyanate-based compn. for coating)

L8 ANSWER 2 OF 10 ZCAPLUS COPYRIGHT 2002 ACS
2000:157971 Document No. 132:212746 Dental adhesive system. Matsunae,
Kaori; Akahane, Shoji (GC Corporation, Japan). Ger. Offen. DE
19942923 A1 20000309, 16 pp. (German). CODEN: GWXXBX.
APPLICATION: DE 1999-19942923 19990908. PRIORITY: JP 1998-254037
19980908.

A dental adhesive system comprises a tooth surface-conditioning AB agent (an acidic aq. soln.) and a binder. The binder is a mixt. of (a) a fluoroaluminosilicate glass powder; (b) .gtoreq.1 unsatd. polymerizable org. compds., either with .gtoreq.1 CH2:CR1CO2 group (R1 = H, Me) or without an acidic group; (c) an acid; (d) water; and (e) a photopolymn. catalyst. The dental prosthetic and the tooth material can be firmly attached with the adhesive in a clin. simple The continuous release of F- from the adhesive assures the restoration of the tooth material, strengthening of the tooth structure, and the inhibition of secondary caries. Thus, a tooth surface-conditioning agent comprised citric acid 15, FeCl3 2, and distd. H2O 83 wt.%. A binder for use therewith was prepd. by mixing 2-hydroxyethyl methacrylate 25, triethylene glycol dimethacrylate 22.5, bis(2-methacryloxyethyl) 2,2,4-trimethylhexamethylenedicarbama te 10, distd. H2O 5, citric acid 5, and fluoroaluminosilicate glass powder 30 wt.% to form a paste which was combined with camphorquinone 1.0 and N,N-dimethylaminoethyl methacrylate 1.5 wt.%. After photopolymn. with visible light, the adhesive adhered to enamel and dentin with a strength of 15.3 and 13.8 MPa, resp.

IT 260391-90-0

(dental adhesive system)

RN 260391-90-0 ZCAPLUS

CN 11,14-Dioxa-2,9-diazaheptadec-16-enoic acid, 4,4,6,16-tetramethyl-10,15-dioxo-, 2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl ester, polymer with 2-hydroxyethyl 2-methyl-2-propenoate, 2-hydroxy-1,3-propanediyl bis(2-methyl-2-propenoate), 2-methyl-2-propenoic acid and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 41137-60-4 CMF C23 H38 N2 O8

PAGE 1-B

CM 2

CRN 1830-78-0 CMF C11 H16 O5

CM 3

CRN 868-77-9 CMF C6 H10 O3

CRN 79-41-4 CMF C4 H6 O2

$$^{\mathrm{CH_2}}_{||}$$
 $^{\mathrm{Me-C-CO_2H}}$

CM 5

CRN 79-10-7 CMF C3 H4 O2

IT 260391-90-0

(dental adhesive system)

L8 ANSWER 3 OF 10 ZCAPLUS COPYRIGHT 2002 ACS
1999:690778 Document No. 131:291351 Radically polymerizable dental
material. Rumphorst, Andre; Salz, Ulrich; Gianasmidis, Alexandros;
Voelkel, Thomas; Moszner, Norbert; Rheinberger, Volker (Ivoclar
A.-G., Liechtenstein). Eur. Pat. Appl. EP 951896 A2 19991027, 10
pp. DESIGNATED STATES: R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT,
LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO. (German). CODEN:
EPXXDW. APPLICATION: EP 1999-250101 19990331. PRIORITY: DE
1998-19818210 19980424.

AB A dental filling or cement compn. contains .gtoreq.1 polymeric binder (polymerizable monomer, oligomer, prepolymer, macromonomer) and .gtoreq.1 filler contg. a 3-part redox initiator system. The latter comprises a homogeneous mixt. of (1) filler mixed with initiator (oxidant), (2) filler mixed with activator (reductant), and (3) filler alone which acts as diluent. The initiator and activator are not microencapsulated as in the prior art, and therefore do not require large mech. forces or long dissoln. times to initiate hardening. These components are premixed in a defined proportion and therefore do not require mixing by the user. Thus, a

statistical oligomer was prepd. by heating acrylic acid 4 and mercaptoethanol 0.4 mol at 95.degree. for 60 min in the presence of azobis (cyanovaleric acid), and treating the product with glycidyl methacrylate 1.6 mol at 60.degree. for 2 days in the presence of hydroguinone mono-Me ether. A binder component was prepd. by mixing this oligomer 24.7, hydroxyethyl methacrylate 27.9, glycerin 1,3-dimethacrylate 27.6, and deionized H2O 19.8 wt.%. A Ca Al fluorosilicate glass powder was silanized with .gamma .methacryloxypropyltrimethoxysilane, and sep. portions were coated with either Bz202 (initiator) or benzylphenylbarbituric acid (activator). A filler mixt. was then prepd. by mixing silanized ionomer glass 59.0, initiator-coated ionomer glass 8.0, activator-coated ionomer glass 6.0, YbF3 25.0, and fumed silica 2.00 The binder and filler components were mixed in a ratio of 1:2.25 and allowed to harden. The hardened product had module of elasticity 3300-3800 MPa, bending strength 40-50 MPa, processing time 120 s at 23.degree., hardening time 360-420 s at 23.degree., and expansion 0.4% after storage for 4 wk in 0.8% NaCl soln. at 37.degree..

IT 246876-42-6P

(binder; radically polymerizable dental material)

RN 246876-42-6 ZCAPLUS

2-Propenoic acid, 2-methyl-, 2-hydroxy-1,3-propanediyl ester, polymer with 2-hydroxyethyl 2-methyl-2-propenoate, 2-mercaptoethanol, oxiranylmethyl 2-methyl-2-propenoate and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CN

CRN 1830-78-0 CMF C11 H16 O5

CM 2

CRN 868-77-9 CMF C6 H10 O3

CRN 106-91-2 CMF C7 H10 O3

CM 4

CRN 79-10-7 CMF C3 H4 O2

CM 5

CRN 60-24-2 CMF C2 H6 O S

 $HO-CH_2-CH_2-SH$

IT 246876-42-6P

(binder; radically polymerizable dental material)

L8 ANSWER 4 OF 10 ZCAPLUS COPYRIGHT 2002 ACS
1989:555138 Document No. 111:155138 Urethane (meth)acrylate resins and
their compositions. Takahashi, Eiji; Morikawa, Takao; Tsuda, Hideo
(Nippon Soda Co., Ltd., Japan). Jpn. Kokai Tokkyo Koho JP 01043516
A2 19890215 Heisei, 12 pp. (Japanese). CODEN: JKXXAF.

APPLICATION: JP 1987-199280 19870810.

Title compns. with excellent soly. and curability, giving coatings with good adhesion, weatherability, hardness, and resistance to acid, base, solvent, and heat, useful for adhesives, inks, resists, etc., comprise A(UR1UX)nUR2UA [R1 = residues of diisocyanates contg. .gtoreq.50 mol% aliph. diisocyanates; R2 = diisocyanate residue; X = diol residues contg. .gtoreq.30 mol% glycerol monomethacrylate or glycerol monoallyl ether; A = residue of (meth)acryloyl-contg. alcs.; U = urethane bond; n = 1-150] with mol. wt. 700-50,000 and

radical polymn. initiators. Thus, 31.12 parts isophorone diisocyanate (I) and 27.75 parts glycerol monoallyl ether (II) were treated in dioxane in the presence of dibutyltin dilaurate at 80.degree. for 3 h to give a urethane prepolymer, which was stirred with 31.12 parts I and 29.99 parts 2-hydroxy-3-methacryloyloxypropyl acrylate to give a title resin (III) with mol. wt. 2300. Then, a glass epoxy substrate was spin coated with a compn. contg. 100 parts of 50% III in dioxane and 2.5 parts benzil di-Me ketal, dried 20 min at 80.degree., irradiated by UV, and postcured at 130.degree. for 30 min to give a coating with pencil hardness 9H, insulating resistance 4.2 .OMEGA. initially and 3.5 .OMEGA. after 65 h at 40.degree. and 95% humidity, and cross-cut adhesion 100/100 after 30 s in solder at 260.degree., 100/100 after 5 min in boiling Triclene, and 100/100after 20 h in Cu plating soln. at 70.degree., vs., 6H, 2.6, 1.5, 0/100, 0/100, and 0/100, resp., for a coating prepd. using cyclohexanedimethanol instead of II.

IT 123088-03-9P 123088-04-0P

i

(preph. of, chem.- and heat-resistant, for coatings and resists) 123088-03-9 ZCAPLUS

2-Propenoic acid, 2-methyl-, 2-hydroxy-3-[(1-oxo-2-propenyl)oxy]propyl ester, polymer with 1,3-bis(isocyanatomethyl)cyclohexane, .alpha.-(2-methyl-1-oxo-2-propenyl)-.omega.-hydroxypoly(oxy-1,2-ethanediyl) and 1,2,3-propanetriol mono(2-methyl-2-propenoate) (9CI) (CA INDEX NAME)

CM 1

RN

CN

CRN 38661-72-2 CMF C10 H14 N2 O2

CM 2

CRN 25736-86-1 CMF (C2 H4 O)n C4 H6 O2 CCI PMS

$$H_2C$$
 O \parallel \parallel \parallel \parallel \parallel OH $Me-C-C$ $CH_2-CH_2-CH_2$ OH

CRN 1709-71-3 CMF C10 H14 O5

CM 4

CRN 50853-28-6 CMF C7 H12 O4

CCI IDS

CM 5

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-- C--- CO}_2\text{H} \end{array}$$

CM 6

CRN 56-81-5 CMF C3 H8 O3

$$^{\rm OH}_{\rm HO-\,CH_2-\,CH-\,CH_2-\,OH}$$

RN 123088-04-0 ZCAPLUS CN 2-Propenoic acid, 2-1

2-Propenoic acid, 2-methyl-, 2-hydroxy-3-[(1-oxo-2-propenyl)oxy]propyl ester, polymer with 1,1'-methylenebis[4-isocyanatocyclohexane], .alpha.-(2-methyl-1-oxo-2-propenyl)-.omega.-hydroxypoly(oxy-1,2-ethanediyl), 1,2,3-propanetriol mono(2-methyl-2-propenoate) and 2(or 3)-(2-propenyloxy)-1,?-propanediol (9CI) (CA INDEX NAME)

CM 1

CRN 25736-86-1 CMF (C2 H4 O)n C4 H6 O2 CCI PMS

$$egin{array}{c|c} H_2C & O \\ \parallel & \parallel & \parallel \\ Me-C-C-C & O-CH_2-CH_2 & ---- OH \\ \hline \end{array}$$
 OH

CM 2

CRN 5124-30-1 CMF C15 H22 N2 O2

CM 3

CRN 1709-71-3 CMF C10 H14 O5

CM 4

CRN 50853-28-6 CMF C7 H12 O4

CCI IDS

CM 5

CRN 79-41-4 CMF C4 H6 O2

CRN 56-81-5 CMF C3 H8 O3

он
$$|$$
 но— ${\rm CH_2}-{\rm CH}-{\rm CH_2}-{\rm OH}$

CM 7

CRN 25136-53-2 CMF C6 H12 O3 CCI IDS

CM 8

CRN 107-18-6 CMF C3 H6 O

 $H_2C = CH - CH_2 - OH$

CM 9

CRN 56-81-5 CMF C3 H8 O3

$$\begin{array}{c} \text{OH} \\ | \\ \text{HO-} \ \text{CH}_2\text{--} \ \text{CH-} \ \text{CH}_2\text{--} \ \text{OH} \end{array}$$

123088-03-9P 123088-04-0P IT

(prepn. of, chem. - and heat-resistant, for coatings and resists)

ANSWER 5 OF 10 ZCAPLUS COPYRIGHT 2002 ACS L8 1988:592294 Document No. 109:192294 Electrodeposition coating

composition containing catalyst-carrying microparticles. Tsuchiya,

Yasuyuki; Nishikawa, Shigeo; Tobinaga, Kenshiro; Muramoto, Hisaichi (Nippon Paint Co., Ltd., Japan). Eur. Pat. Appl. EP 264834 Al 19880427, 18 pp. DESIGNATED STATES: R: DE, FR, GB, IT. (English). CODEN: EPXXDW. APPLICATION: EP 1987-115100 19871015. PRIORITY: JP 1986-246770 19861016; JP 1986-246771 19861016.

The title compn., providing coating films with good hardness, corrosion and solvent resistance, comprises aq. dispersions of film-forming base resins comprising oxidatively polymerizable resins and/or mixts. of active H-contg. polymers and blocked polyisocyanates, and dispersion-dispersible polymer microparticles carrying catalytically effective amts. of metal compds. or tertiary Thus, a varnish [prepd. from Nisseki B 1500 amines on the surface. (polybutadiene), Antigene 6C (N-methyl-N'-(1,3-dimethylbutyl)-pphenylenediamine), maleic anhydride, Et2NH, and HO(CH2)3OH] 125, another varnish [prepd. from Epo Tohto YD 014 (epoxy resin), acrylic acid, and hydroquinone] 75, 50% (solids) melamine resin 40, 50% phenolic resin 40, and nonionic surfactant 2 parts were mixed, stirred with a mixt. of 13 parts Et3N, 707 parts H2O, and 60 parts microparticles prepd. from (methacryloyloxy)zirconium octoate, styrene, N-methyl-N-(vinylbenzyl)taurine, 2-hydroxyethyl acrylate, Me methacrylate, Bu acrylate, monobutyl maleate-glycidyl methacrylate adduct, Bu methacrylate, 2-hydroxyethyl methacrylate, acrylic acid, and ethylene glycol dimethacrylate. The mixt. was then elec. deposited on a Zn phosphate-treated steel plate, washed, and baked at 140.degree. for 0.5 h to give a coating showing pencil hardness HB, no change after rubbing 20 times with a gauze fabric impregnated with iso-BuCOMe, and low corrosion (240 h, 5% saline spraying); vs. 2B, poor solvent resistance, and high corrosion without the microparticles.

IT 117316-80-0

RN

CN

AB

(microparticles, elec. deposited coatings contg., for good resistance to corrosion and rubbing)

117316-80-0 ZCAPLUS

Zirconium, (2-methyl-2-propenoato-0) (octanoato-0)-, polymer with (Z)-butyl 2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl 2-butenoate, butyl 2-methyl-2-propenoate, 1,2-ethanediyl bis(2-methyl-2-propenoate), ethenylbenzene, 2-[(ethenylphenyl)methylamino]ethanesulfonic acid, 2-hydroxyethyl 2-methyl-2-propenoate, 2-hydroxyethyl 2-propenoate, 2-methoxyethanol, methyl 2-methyl-2-propenoate and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 117316-79-7 CMF C11 H15 N O3 S CCI IDS



$$D1-CH=CH_2$$

$$\begin{array}{c} \text{D1} \\ | \\ \text{Me-N-CH}_2\text{--CH}_2\text{--SO}_3\text{H} \end{array}$$

CRN 114571-40-3 CMF C12 H20 O4 Zr

CM 3

CRN 113149-99-8 CMF C15 H22 O7

Double bond geometry as shown.

CM 4

CRN 868-77-9 CMF C6 H10 O3

CRN 818-61-1 CMF C5 H8 O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{--CH}_2\text{--O-C--CH} \\ \end{array}$$

CM 6

CRN 109-86-4 CMF C3 H8 O2

CM 7

CRN 100-42-5 CMF C8 H8

$$H_2C = CH - Ph$$

CM 8

CRN 97-90-5 CMF C10 H14 O4

CRN 97-88-1 CMF C8 H14 O2

CM 10

CRN 80-62-6 CMF C5 H8 O2

$$^{\mathrm{H_2C}}$$
 O $^{\mathrm{H_2C}}$ Me- C- C- OMe

CM 11

CRN 79-10-7 CMF C3 H4 O2

IT 117316-80-0

(microparticles, elec. deposited coatings contg., for good resistance to corrosion and rubbing)

L8 ANSWER 6 OF 10 ZCAPLUS COPYRIGHT 2002 ACS

1988:114298 Document No. 108:114298 Core-shell composite acrylic resin particles. Miyazono, Tadafumi; Kashihara, Akio; Ishikura, Shinichi (Nippon Paint Co., Ltd., Japan). Eur. Pat. Appl. EP 242235 A2 19871021, 14 pp. DESIGNATED STATES: R: DE, FR, GB. (English). CODEN: EPXXDW. APPLICATION: EP 1987-303493 19870421. PRIORITY: JP 1986-90827 19860418.

AB Title resin particles useful in paints and coatings, comprising particulate crosslinked acrylic polymer with substantially chem. bonded linear acrylic polymer chains, are prepd. by emulsion polymg. compds. having .gtoreq.2 mono- and/or 1,1-disubstituted ethylenically unsatd. groups, monofunctional polymerizable non-arom.

monomers, and 1,2-di, 1,1,2-tri-, 1,1,2,2-tetrasubstituted ethylenically unsatd. compds. to obtain particles of a polymer contg. unsatd. groups, and graft polymg. with arom. compds. and optionally other monofunctional monomers. Thus, 430 parts Bu maleate was heated to 150.degree. in the presence of hydroguinone, and mixed with 373 parts glycidyl methacrylate for 90 min to give a mixt., 12 parts of which was mixed with Me methacrylate 64, Bu acrylate 20, and ethylene glycol dimethacrylate 40 parts. Adding the acrylic mixt. to a dispersion contg. amphoteric polyester stabilizer 20, dimethanolamine 2, and water 104 parts gave an emulsion, which was then added dropwise to 330 parts water at 80.degree. over 60 min together with initiator soln. Allowing the mixt. to stand at 80.degree. for 30 min, then slowly adding a mixt. of styrene 16, Bu acrylate 8, and Me methacrylate 16, and initiators 0.8 part gave crosslinked polymer particles readily dispersible in xylene, BuOAc, MEK, ethylene glycol, and Bu ether, forming 30%-solids compns. with viscosities 250, 230, 220, and 260 P, resp. Applying the org. solvent dispersions to glass plates and drying gave clear coatings.

IT 113149-98-7P

(core-shell composite particles, manuf. of, with good dispersion stability, for coatings)

RN 113149-98-7 ZCAPLUS

CN 2-Butenedioic acid, 2,3-dimethyl-, butyl 2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl ester, (Z)-, polymer with butyl 2-propenoate, ethenylbenzene, 1,6-hexanediyl bis(2-methyl-2-propenoate), 2-hydroxyethyl 2-methyl-2-propenoate, methyl 2-methyl-2-propenoate and 2-methyl-2-propenoic acid, graft (9CI) (CA INDEX NAME)

CM 1

CRN 113149-97-6 CMF C17 H26 O7

Double bond geometry as shown.

CM 2

CRN 6606-59-3 CMF C14 H22 O4

$$^{\rm H_2C}$$
 O $^{\rm O}$ CH2 $_{\rm \parallel}$ $_{\rm \parallel}$ $_{\rm \parallel}$ Me-C-C-O-(CH2)6-O-C-C-Me

CRN 868-77-9 CMF C6 H10 O3

$$^{\mathrm{H_{2}C}}_{\parallel}$$
 $^{\mathrm{C}}_{\parallel}$ $^{\mathrm{C}}_{\parallel}$ $^{\mathrm{CH_{2}-CH_{2}-CH_{2}-OH}}$

CM 4

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{n-BuO-C-CH} \stackrel{\longleftarrow}{\longleftarrow} \text{CH}_2 \end{array}$$

CM 5

CRN 100-42-5 CMF C8 H8

$${\tt H_2C} \overline{=\!\!\!\!=} {\tt CH-Ph}$$

CM 6

CRN 80-62-6 CMF C5 H8 O2

CRN 79-41-4 CMF C4 H6 O2

CH₂ || Me-C-CO₂H

IT 113149-98-7P

(core-shell composite particles, manuf. of, with good dispersion stability, for coatings)

L8 ANSWER 7 OF 10 ZCAPLUS COPYRIGHT 2002 ACS

1978:90438 Document No. 88:90438 Adhesives for anaerobic hardening.
Gruber, Werner; Galinke, Joachim; Wegemund, Bernd (Henkel und Cie.
G.m.b.H., Ger.). Ger. Offen. DE 2607961 19770901, 13 pp. (German).
CODEN: GWXXBX. APPLICATION: DE 1976-2607961 19760227.

Bis(hydroxyethyl) carbonate (I) [55163-79-6] or a mixt. of I and AB HOCH2CH2OH [107-21-1] was esterified (1 mol) with two mol maleic anhydride (II) [108-31-6] and/or phthalic anhydride [85-44-9] to prep. dicarboxy compds. which were esterified with glycidyl methacrylate (III) [106-91-2] to prep. dimethacrylates. dimethacrylates were used with monomers such as hydroxyethyl methacrylate (IV) [868-77-9] and methacrylic acid (V) [79-41-4] to prep. adhesives which hardened anaerobically and were used for bonding metal sheets. Thus, 0.2 mol I was esterified with 0.4 mol II and treated with 50.5 g III to prep. a dimethacrylate ester [64461-12-7] which was used (70 g) with IV 20, V 3, p-toluenesulfonyl hydrazide 1, N, N-dimethyl-p-toluidine 1, 70% cumene hydroperoxide in cumene 5, and 40% AcOOH in AcOH 1 g to prep. an adhesive which hardened between steel sheets to give tensile shear strength 245 kg/cm2.

IT 64461-13-8

(adhesives, for anaerobic hardening)

RN 64461-13-8 ZCAPLUS

CN 5,8,10,13-Tetraoxaheptadeca-2,15-dienedioic acid, 4,9,14-trioxo-, bis[2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl] ester, (Z,Z)-, polymer with 2-hydroxyethyl 2-methyl-2-propenoate and 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 64461-12-7 CMF C27 H34 O17

Double bond geometry as shown.

PAGE 1-A

PAGE 1-B

$$\begin{array}{c|c} & \text{OH} & \text{CH}_2 \\ \hline \hline z & \\ \hline O & O & O \end{array}$$

CM 2

CRN 868-77-9 CMF C6 H10 O3

$$^{\rm H_2C}$$
 O $^{\parallel}$ $^{\parallel}$ $^{\parallel}$ Me-C-C-O-CH₂-CH₂-OH

CM 3

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

IT 64461-13-8

(adhesives, for anaerobic hardening)

L8 ANSWER 8 OF 10 ZCAPLUS COPYRIGHT 2002 ACS
1978:7874 Document No. 88:7874 Adhesives for anaerobic hardening.
Gruber, Werner; Galinke, Joachim (Henkel und Cie. G.m.b.H., Fed.

Rep. Ger.). Ger. Offen. DE 2607960 19770901, 13 pp. CODEN: GWXXBX. APPLICATION: DE 1976-2607960 19760227. [32158-14-8] was esterified (1 mol) with 2 AB EtO2CN(CH2CH2OH)2 (I) mols maleic anhydride (II) [108-31-6], phthalic anhydride [85-44-9]-II mixt., or cyclohexanedicarboxylic anhydride to prep. carboxy-terminated esters which were esterified with qlycidyl [106-91-2] to prep. dimethacrylates. methacrylate (III) dimethacrylates were used with methacrylic acid (IV) [79-41-4] to bond metals. Thus, 106 g I was esterified with 117 g II to prep. a carboxy-terminated ester which was treated with 48.6 g III to prep. a dimethacrylate ester [64462-31-3]. A mixt. of the dimethacrylate ester 70, hydroxyethyl methacrylate [868-77-9] 20, N, N-dimethyl-p-toluidine 0.5, 70% cumene hydroperoxide in cumene 5, 40% AcooH in AcoH 1, and IV 3.5 g was used to bond steel sheets, giving tensile shear strength 235 kg/cm2.

IT 64502-57-4

CN

(adhesives, for metals, for anaerobic hardening)

RN 64502-57-4 ZCAPLUS

> 2-Butenedioic acid (2Z)-, [(ethoxycarbonyl)imino]di-2,1-ethanediyl bis[2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl] ester, polymer with 2-hydroxyethyl 2-methyl-2-propenoate and 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 64462-31-3 CMF C29 H39 N O16

Double bond geometry as shown.

PAGE 1-B

$$\overline{z}$$
 OH CH_2 Me

CRN 868-77-9 CMF C6 H10 O3

CM 3

CRN 79-41-4 CMF C4 H6 O2

$$^{\mathrm{CH_2}}_{||}$$
 Me-C-CO₂H

IT

IT 64502-57-4

(adhesives, for metals, for anaerobic hardening)

L8 ANSWER 9 OF 10 ZCAPLUS COPYRIGHT 2002 ACS
1977:568778 Document No. 87:168778 Adhesives for anaerobic hardening.
Wegemund, Bernd; Gruber, Werner; Galinke, Joachim (Henkel und Cie.
G.m.b.H., Ger.). Ger. Offen. DE 2607962 19770901, 15 pp. (German)
CODEN: GWXXBX. APPLICATION: DE 1976-2607962 19760227.

HOCH2CH2OH [107-21-1], 1,4-butanediol [110-63-4], a mixt. of AB1,4-butanediol and 2-butyne-1,4-diol [110-65-6], or similar diols were esterified (1 mol) with 2 mol maleic anhydride (I) [108-31-6] phthalic anhydride [85-44-9]-I mixt., or cyclohexanedicarboxylic anhydride [85-42-7], and the dicarboxy product was esterified with glycidyl methacrylate (II) [106-91-2] to prep. dimethacrylates which were used with monomers such as hydroxyethyl methacrylate (III) [868-77-9] and methacrylic acid (IV) [79-41-4] to prep. adhesives suitable for anaerobic hardening. The adhesives were esp. useful for bonding metals and gave bonds with good strength, heat stability, and flexibility. Thus, 156.8 g I was estified with 49.6 q HOCH2CH2OH at 85.degree. and esterified further with 234 g II at 80.degree. to prep. a dimethacrylate ester [53562-93-9] which was mixed (70 g) with a methacrylate ester of 5,6dihydrodicyclopentadienol 10, III 10, p-toluenesulfonyl hydrazide 0.5, N,N-dimethyl-p-toluidine 0.5, 70% soln. of cumene hydroperoxide in cumene 5, 40% soln. of AcOOH in AcOH 1, and IV 3 parts and used

to bond steel sheets with tensile shear strength 270 kg/cm2.

64519-10-4 64520-75-8

(adhesives, for anaerobic hardening)

RN 64519-10-4 ZCAPLUS

2-Butenedioic acid (2Z)-, 1,4-butanediyl bis[2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl] ester, polymer with 2-hydroxyethyl 2-methyl-2-propenoate and 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 55554-71-7 CMF C26 H34 O14

PAGE 1-A

PAGE 1-B

CM 2

CRN 868-77-9 CMF C6 H10 O3

$$^{\mathrm{H_{2}C}}_{\parallel}$$
 $^{\mathrm{O}}_{\parallel}$ $^{\mathrm{Me-C-C-O-CH_{2}-CH_{2}-OH}}$

CM 3

CRN 79-41-4 CMF C4 H6 O2

$$^{\mathrm{CH_2}}_{||}$$
 Me $^-$ C $^-$ CO $_2$ H

RN 64520-75-8 ZCAPLUS

CN 2-Butenedioic acid (2Z)-, 1,2-ethanediyl bis[2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl] ester, polymer with 3a,4,5,6,7,7a-hexahydro-4,7-methano-1H-inden-5(or 6)-yl 2-methyl-2-propenoate, 2-hydroxyethyl 2-methyl-2-propenoate and 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 53562-93-9 CMF C24 H30 O14

PAGE 1-B

CM 2

CRN 31621-69-9 CMF C14 H18 O2 CCI IDS

CM 3

CRN 868-77-9 CMF C6 H10 O3

CM 4

CRN 79-41-4 CMF C4 H6 O2

IT 64519-10-4 64520-75-8

(adhesives, for anaerobic hardening)

L8 ANSWER 10 OF 10 ZCAPLUS COPYRIGHT 2002 ACS

1975:87806 Document No. 82:87806 Entropically stabilized water base dispersions. Parker, Fred Walter (du Pont de Nemours, E. I., and Co.). U.S. US 3801524 19740402, 6 pp. Continuation-in-part of U.S. 3,729,439 (CA 79;19669c). (English). CODEN: USXXAM. APPLICATION: US 1972-316405 19721218.

Ag. dispersions which were useful as finishes for automobile and AB truck bodies were prepd. from graft copolymers having graft segments which contain hydroxy-terminated acrylic monomers. Thus, an aq. dispersion of allyl methacrylate-butyl methacrylate-methyl methacrylate-acrylic acid-2-hydroxyethyl acrylate graft copolymer [40738-48-5] was prepd. by refluxing a mixt. contg. Me methacrylate 700, Bu methacrylate 140, allyl methacrylate 10, anhyd. iso-PrOH (I) 350, BuO(CH2)2OH 75, and BuOH (II) 117 parts, adding a mixt. contg. Vazo initiator (azobisisobutyronitrile) 3, I 12, and II 3 parts and refluxing, adding a mixt. contg. Vazo initiator 1.5, I 7, and II 3 parts and cooling the resultant mixt. to 87.degree., adding a mixt. 2-hydroxyethyl acrylate 103, acrylic acid 50, I 49, II 16, and Lupersol 11 (75% soln. of tert-Bu peroxypivalate in mineral spirits) (III) 1.5 parts, heating the resultant mixt. 1 hr at 87.degree., adding 15 parts I and 0.8 part III, cooling to 77.degree., and adding 15 parts I and 0.4 part III followed by 134 parts I and 500 parts demineralized H2O.

IT 52409-04-8

(graft, in aq. coatings)

RN 52409-04-8 ZCAPLUS

CN 2-Propenoic acid, 2-methyl-, butyl ester, polymer with

2-hydroxyethyl 2-propenoate, 2-hydroxy-3-[(1-oxo-2-propenyl)oxy]propyl 2-methyl-2-propenoate, methyl 2-methyl-2-propenoate and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 1709-71-3 CMF C10 H14 O5

CM 2

CRN 818-61-1 CMF C5 H8 O3

CM 3

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 4

CRN 80-62-6 CMF C5 H8 O2

CRN 79-10-7 CMF C3 H4 O2

$$\begin{matrix} \text{O} \\ || \\ \text{HO-C-CH} = \text{CH}_2 \end{matrix}$$

IT **52409-04-8** (graft, in aq. coatings)